Lab Instructor Information

Instructor: ________________________________________, e-mail ____________________, Sec _____

Lab instructors are teaching assistants (TAs) selected by the Department of Chemistry and are well educated in the field of chemistry. In addition to giving you an introduction to each experiment, your instructor is responsible for overseeing your safety while in the lab and for assisting you in your lab work. Your instructor can be your biggest asset in this course. Don’t hesitate to ask your instructor questions, however s/he is not a substitute for preparation and will not do your experiment for you or give you answers to complete your report. Please record your instructor’s contact information at your first meeting. Instructor assignments are on page 2.

Lab Supervisor Information

Supervisor: Lorri Reilly, MS
Office Location: 508B Chem Sci
Telephone: 906-487-2044
E-mail: lareilly@mtu.edu

To report an absence: E-mail absentfyclab@mtu.edu, see the Absences & Makeups policy in this syllabus for detailed information.

The lab supervisor is responsible for course design and administration and assists instructors and students as needed. Please feel free to contact her if you have an issue that you don’t wish to discuss with your instructor.

Course Office Hours

Office hours for CH1151 are held in Rm 508 Chem Sci and are scheduled prior to the start of each lab session; Tues & Thurs 11 am and 2 pm; Tues at 6:30 pm. Feel free to stop by, especially if you need assistance completing your prelab assignment.

Course Identification

Course Number: CH1151 L01-L09 (see section assignments on page 2)
Course Name: University Chemistry I Lab
Course Location: 5th floor Chem Sci (see room assignments on page 2)
Class Times: (see section assignments on page 2)
Co-requisite: CH1150
Pre-requisite: Grade ≥ C in one of the following: MA 1031, MA1032, MA1135, MA 1160, or MA 1161 OR ALEKS Math Placement ≥ 56 OR CEEB Calculus (AB, AB Subscore, or BC) ≥ 2

Course Description/Overview

CH1151 is the laboratory component that emphasizes CH1150 lecture material (introduces the foundations of chemistry, including electronic structure of atoms and molecules, intermolecular forces, states of matter, chemical reactions, acid-base chemistry, kinetics, chemical equilibria, and some organic chemistry).

Course Learning Objectives

Chemistry is a laboratory science and experimentation is the foundation of chemical knowledge. Experiments are chosen to emphasize material covered in CH1150 lecture and the lab is designed to provide you with
practical experience in conducting tests and making observations. As you progress through this course it is expected that you will develop and enhance the skills needed to collect meaningful data, interpret the results, draw conclusions, and communicate your findings.

CH1151 also supports students in achieving University Student Learning Goal #2: Knowledge of the Physical and Natural World. Specific student competencies for this goal can be found under the link for Goal #2, see: http://www.mtu.edu/assessment/program/university-learning-goals/

### Section/Room/Instructor Assignments/Spring 2016

<table>
<thead>
<tr>
<th>TUESDAY</th>
<th>THURSDAY</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>noon - 3 pm</strong></td>
<td><strong>noon - 3 pm</strong></td>
</tr>
<tr>
<td>CH1151 L01 503N Zach Otis</td>
<td>CH1151 L06 503N Closed</td>
</tr>
<tr>
<td>CH1151 L02 503S Jacob Pleiness</td>
<td>CH1151 L07 503S Jacob Pleiness</td>
</tr>
<tr>
<td><strong>3 - 6 pm</strong></td>
<td><strong>3 - 6 pm</strong></td>
</tr>
<tr>
<td>CH1151 L03 503N Ashley Schuman</td>
<td>CH1151 L08 503N Christina Welch</td>
</tr>
<tr>
<td>CH1151 L04 503S Jacquelyn Hod</td>
<td>CH1151 L09 503S Jacquelyn Hood</td>
</tr>
<tr>
<td><strong>7 - 10 pm</strong></td>
<td></td>
</tr>
<tr>
<td>CH1151 L05 503S Joe Vermeylen</td>
<td></td>
</tr>
<tr>
<td>CH1151 Makeups 503N Braden LaNore*</td>
<td></td>
</tr>
</tbody>
</table>

*This makeup session is for the previous week’s experiment. Approved makeup permission slips must be picked up from Rm 508 when you arrive for the makeup.

### Course Resources

**Canvas CH1151 site:** Access to lab grades, experimental handouts, prelabs, & other course information/materials


**Course Fees:** $223; includes lab fee, experimental handouts, eye protection

$25 fine for failure to check out drawer equipment by 5 pm Thurs of week 14

See page 8 for costs associated with missing or broken equipment

**Course Supplies:** Proper attire required - Wear clothing that provides the most protection. Shorts, skirts or dresses must be knee length or longer. Open shoes (such as sandals) are not allowed, even with socks underneath. Students dressed inappropriately will be sent home to change. Tie long hair back.

**Calculator & Pen:** Bring them with you to lab.

**Binder:** Recommended
## Grading Scheme*

<table>
<thead>
<tr>
<th>Grade</th>
<th>Percentage</th>
<th>Grade pts/credit</th>
<th>Rating</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>≥ 90%</td>
<td>4.00</td>
<td>Excellent</td>
</tr>
<tr>
<td>AB</td>
<td>85% – 89%</td>
<td>3.50</td>
<td>Very good</td>
</tr>
<tr>
<td>B</td>
<td>80% – 84%</td>
<td>3.00</td>
<td>Good</td>
</tr>
<tr>
<td>BC</td>
<td>75% – 79%</td>
<td>2.50</td>
<td>Above average</td>
</tr>
<tr>
<td>C</td>
<td>70% – 74%</td>
<td>2.00</td>
<td>Average</td>
</tr>
<tr>
<td>CD</td>
<td>65% – 69%</td>
<td>1.50</td>
<td>Below average</td>
</tr>
<tr>
<td>D</td>
<td>60% - 64%</td>
<td>1.00</td>
<td>Inferior</td>
</tr>
<tr>
<td>F</td>
<td>≤59%</td>
<td>0.00</td>
<td>Failure</td>
</tr>
<tr>
<td>I</td>
<td>Incomplete</td>
<td></td>
<td></td>
</tr>
<tr>
<td>X</td>
<td>Conditional</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

*NOTE: The percentage calculated by Canvas may not reflect your actual percentage in lab. To determine your actual percentage follow the grading policy detailed below.

### Grading Policy 100 lab points total, see following point distribution:

- **Experimental** (7 of the experiments will be graded @12 pts each, which 7 is unknown to you, lowest score dropped) 72 pts
- **Quizzes** (2 @ 14 pts each, graded on correctness) 28 pts

**Experimental**: You are expected to attend every lab session and complete all experiments. Every report will be collected and reviewed, however only seven will be graded. You will not know which seven. Each is worth 12 points and is graded as described below. Your lowest experimental score will be dropped. For each non-graded experiment missed and not made up, 5 points will be deducted from your total points at the end of the semester. See the following page for information regarding absences/makeups.

- **Three of twelve points for the following:**
  - **Preparation for each lab** (coming to lab on time, with approved eye protection and drawer combination, properly dressed, and prelab assignments completed) Note that prelabs are due when you enter the lab. You will lose points if your prelab is submitted after your instructor begins her/his introduction.
  - **Effort during lab** (knowledge and understanding of the procedure and chemistry involved; organization and ability to budget time, common equipment check in/out procedures followed, cooperation and attitude)
  - **Safety rules followed** (approved eye protection worn at all required times, proper waste disposal, clean work areas, labware cleaned at completion of experiment)

- **Six of twelve points for completion of the report sheet** (data and observations recorded correctly and comprehensively, one example of each calculation performed shown in detail,
appropriate graphs; values shown in data and calculations have the appropriate number of significant figures, units and labels, lead zeroes reported; data recorded in ink, one single line through mistakes). Grading criteria for these points differ for each experiment and may include points for obtaining accurate and precise results.

- **Three of twelve points for completion of the postlab assignment** (postlab assignments will be distributed in lab; your answers must be clear, concise and comprehensive)

**Lab Quizzes:** Two quizzes are scheduled, each worth 14 pts. See the experiment schedule on the last page of this syllabus for dates. All work must be shown to receive credit. You will need a calculator and may use a pencil on quizzes. Dedicated language translators are permitted. Cell phones are not permitted.

**Course Policies**

**ABSENCES & MAKEUPS:** Missed experiments must be made up within a specified time frame (see below) if you wish to recover the points. Also note that you are responsible for understanding the material covered in the experiment for laboratory quizzes and that the quiz may precede the day of your makeup session. **Makeups can ONLY be performed either the week the experiment is scheduled during one of the time slots shown on page 2 of this syllabus or during the last makeup session held the following week on Tuesday at 7 pm.**

If your circumstances justify additional consideration you should contact your lab supervisor, Lorri Reilly, and explain why. Exceptions may require consultation with the Dean of Students. In the event of a university sponsored activity, students should report the anticipated absence before the missed lab session. In case of illness, absences should be reported as soon as it is known that the lab session will be missed. It is in your best interest to complete your makeup as soon as possible. Note that there are no experimental makeups during week 14 so the week 13 experiment will need to be made up during week 13. Don’t make absence requests a habit!

**HOW TO REQUEST A MAKEUP:** Email absentfyclab@mtu.edu IMMEDIATELY to report your absence. Include the reason for your absence (honesty rewarded). You’ll receive an auto reply message with a Makeup Request Form and the following information:

1. Read the information that follows before replying to this email.
2. Complete the **Makeup Request Form** at the bottom of the email with care. Return your completed form via the absentfyclab email account. Please do not separate or delete portions of the makeup request form.
3. You may assume your request has been granted unless you hear otherwise. You do not need to do anything else other than show up for your makeup session prepared and on time.
4. Stop by Rm 508 B when you arrive for the makeup. You will receive an approved copy of your completed makeup request form that will include the makeup instructor’s name and the room location for your makeup. Submit the form and your completed prelab assignment to the makeup lab instructor so that s/he will know you have permission to be there. Before leaving the lab, submit the rest of your report to the makeup instructor.

You will need the information that follows to complete the **Makeup Request Form** (incomplete request forms will be returned to you and delay scheduling of the makeup):
• Course # (CH1151 or CH1161)
• Section # or Day and Time of regular session (see page 2 of this syllabus for assistance)
• Your instructor’s name (also available on page 2)
• Name of experiment that you need to make up (see last page of syllabus)
• Date, Day and Time you plan to make up (example: 30 Sept Tues at noon)

SAFETY: Chemistry as an experimental science is an exciting subject; however, experimentation in a chemistry lab has an element of danger and risk associated with it. This is particularly true if the surroundings, tools, or techniques are unfamiliar to you. It is important to be prepared and to THINK SAFETY- both for you and others in the lab. If in doubt about the safety of any procedure, ASK the instructor before proceeding. To keep the risk to a minimum, you are expected to adhere to the following rules, regulations and safe techniques. This information applies to students in all chemistry teaching labs.

RULES, REGULATIONS AND SAFE TECHNIQUES

1. Do not enter the teaching laboratory until an instructor or supervisor is present. Guests are not allowed in the lab unless approved by the lab supervisor. The Prep Room, Rm. 506, is off limits to students enrolled in CH1151 and CH1161.

2. On the first day of lab, determine the location and use of all emergency and safety equipment: safety shower, eye wash and fire extinguisher. Ask your instructor if you are uncertain. Locate the nearest emergency exit. Know the emergency evacuation route (posted on the lab door).

3. Wear approved eye protection at ALL times while in the lab, even during clean up. Contact lenses are not recommended as chemicals may become trapped underneath them.

4. Wear clothing that provides the most protection - a lab coat or apron is recommended. Shorts, skirts or dresses must be knee length or longer. Open shoes (such as sandals) are not allowed. Confine long hair and loose clothing when working. Do not wear your favorite clothing to lab.

5. Food, beverages, chewing tobacco and electronic cigarettes are prohibited and must not be visible in the lab (keep them in your book bag). Never eat, drink or taste anything (food, chewing gum, beverages, chemicals, etc...) while in the lab. Do not place fingers, pencils, pipets, etc. in your mouth, nose, or ears. Don’t rub your eyes.

6. Do only the assigned experiment.

7. In case of chemical contact with skin or eyes, flush the affected area with running water for 15 minutes. Use faucets, safety showers or eye wash, as necessary. Remove all contaminated clothing immediately.

8. Report all accidents or injuries to your instructor immediately, even seemingly minor ones.

9. Always avoid unnecessary hazards. Keep working surfaces clean at ALL times. Do not sit or lean on bench surfaces. Keep the floor clear of tripping hazards. Jackets and bookbags should be stored on the coat racks. Stools and chairs are not permitted in the aisles. Drawers should remain closed except when removing equipment.

10. Read the labels on reagent bottles to make sure you have the proper reagents. (Report empty reagent bottles to your instructor.)
11. Do not return excess reagents to stock containers; share with others or dispose of it.

12. Dispose of waste properly. This means:
   a. Broken glass - glass disposal container
   b. Water-soluble liquids - flush down sink
   c. Paper products - waste basket
   d. Solid wastes and water-insoluble waste - properly labeled waste containers. Do not throw solid materials into sinks.
   e. If you are uncertain of the proper waste disposal, consult your instructor.

13. Report all chemical spills immediately and clean up the spill as directed by your instructor. Acid spills should be neutralized with solid sodium bicarbonate (baking soda) or calcium carbonate and base spills should be neutralized using citric acid before wiping them up with a damp sponge. These supplies are available in the lab.

14. Do not test odors by direct inhalation from the container. Fan the vapors gently towards your nose.

15. Always add concentrated acid to water and acids to bases (add an acid to a base if you want to save your face). Pour slowly while stirring the mixture constantly, otherwise it may splatter.

16. Always use a suction bulb (never your mouth) when filling a pipet.

17. Do not insert pipets directly into reagent bottles as the reagent may become contaminated. Transfer an approximate amount into a beaker or other container. The opening on the receiving container should be larger than the opening on the container it is transferred from.

18. Do not force glass tubing and/or thermometers into rubber stoppers. Always lubricate the hole in the stopper with glycerin or soapy water and protect your hand with a towel when inserting tubing or thermometers.

19. Never use an open flame (such as a gas burner or match) in the vicinity of flammable materials and never leave a lighted burner unattended.

20. Return all equipment clean and to its proper location. Do not put common equipment in your drawer. Never remove chemicals from the lab.

21. If protective gloves are issued, remove contaminated gloves immediately after use and wash your hands. Never leave the lab or touch common surfaces (faucets, door knobs, etc...) with contaminated gloves.

22. Wash your hands and arms thoroughly before leaving the lab.

Remember that simple tasks, often regarded as safe, can be dangerous if done improperly. The majority of accidents reported in the laboratory involve cuts from handling broken or chipped glassware and burns from touching hot objects. Always use good judgment and care when working in the lab.

**HAZARD COMMUNICATION STANDARD**

By signing the *Laboratory Worker Safety Agreement* on your check in card, you are stating that you are aware that you have a "right to know" all safety information contained in the manufacturers Safety
Data Sheet (SDS) for any chemical. You can obtain this information by requesting a copy of the SDS from the lab supervisor or from Chem Stores, in Room B002, Chemical Sciences Building (in basement).

**EMERGENCY BUILDING EVACUATION PROCEDURE**

1. The signal to evacuate the building in case of emergency is the building fire alarm. Prepare to evacuate the building if you hear the alarm. Shut off hot plates, turn off lights and electricity, and close windows and doors (to prevent the spread of smoke and fire) before departing.

2. Leave the building in an orderly manner via the evacuation route posted on the inside of the door to the laboratory. Walk, do not run or push. Do not use the elevators. Elevators are only for use by rescue personnel during a building evacuation. The building attendant brings the elevators to the first floor once the alarm is activated and anyone inside could become trapped.

3. Once outside, meet with your instructor at your designated meeting place. This location is different for each of the laboratories and is announced during the first lab session (this location is also indicated on the inside of the lab door). If you do not meet your instructor, s/he will assume that you are trapped in the lab and send rescue workers into the building to look for you. You must stay 100 feet away from the building to enable rescue personnel to get to the site of the emergency.

4. Do not re-enter the building until the all-clear is given by Public Safety.

**DRAWER EQUIPMENT:** At check in you will be assigned a drawer containing some of the equipment that you will use, and be responsible for, during the semester. The drawer contents should be inspected and compared to the equipment list that follows to determine that all items are present and in acceptable condition. If you are uncertain of the identity of any piece of equipment on the list, see the illustrations posted on the bulletin board in the lab. If an item is broken or missing, list it on the yellow equipment replacement form (available from your instructor). After you have finished checking all of the equipment, bring the equipment replacement form to your instructor for replacement glassware. **No replacements will be made after the first day of class.** Sign and return the check in card when you have all the equipment. Once you sign the check in card you are responsible for this equipment and will be charged for any breakage or loss. **You must check out of your drawer by 5 pm Thursday of week 14, even if you drop the course.** Failure to check out results in a $25 fine in addition to charges for missing or broken equipment. Every item on the drawer equipment list must be clean and in the drawer in order to check out. Replacement equipment can be purchased from Chem Stores, Room B002, in the basement of the Chemical Science and Engineering Building. Prices on the list are approximate and subject to change without notice and also include Michigan’s 6% sales tax. Please note that Chem stores does not accept credit or debit cards.

**COMMON EQUIPMENT:** In addition to drawer equipment, you will also use common equipment. Common equipment is equipment also used by other students. Common equipment must be signed out (list both your first and last name). Before doing so, inspect it to ensure that it is in satisfactory condition. Do not sign it out if there is something wrong with it, instead notify your instructor. Once you have accepted the equipment, you are responsible for returning it in clean and satisfactory condition. Before returning the common equipment to its proper location, bring it to your instructor for inspection. Your instructor will initial its return if it is found to be acceptable. Broken or misplaced common equipment should be reported to your instructor immediately. You are responsible for the
cost of replacing such equipment and will receive equipment replacement instructions the following week. Lab fees cover the cost of expendable items such as chemicals, not breakage.

**DRAWER EQUIPMENT LIST** (Prices listed are approximate, include Michigan’s 6% sales tax, and are subject to change without notice.)

<table>
<thead>
<tr>
<th>Quantity</th>
<th>Drawer Equipment</th>
<th>Stock #</th>
<th>Cost, Each (as of May 2015)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Beaker, 50 mL, Pyrex</td>
<td>00537</td>
<td>$2.30</td>
</tr>
<tr>
<td>1</td>
<td>Beaker, 100 mL, Pyrex</td>
<td>00538</td>
<td>$1.91</td>
</tr>
<tr>
<td>2</td>
<td>Beaker, 150 mL, Pyrex</td>
<td>00539</td>
<td>$2.54</td>
</tr>
<tr>
<td>1</td>
<td>Beaker, 250 or 400 mL, Pyrex</td>
<td>00540 or 541</td>
<td>$5.24 or 2.52</td>
</tr>
<tr>
<td>1</td>
<td>Beaker, any size, Pyrex</td>
<td></td>
<td></td>
</tr>
<tr>
<td>1</td>
<td>Erlenmeyer flask, 50 mL</td>
<td>00791</td>
<td>$3.07</td>
</tr>
<tr>
<td>1</td>
<td>Erlenmeyer flask, 125 mL</td>
<td>00792</td>
<td>$3.75</td>
</tr>
<tr>
<td>2</td>
<td>Erlenmeyer flask, 250 mL</td>
<td>00793</td>
<td>$4.03</td>
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<tr>
<td>1</td>
<td>Graduated cylinder, 10 mL</td>
<td>00685</td>
<td>$5.75</td>
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<td>1</td>
<td>Graduated cylinder, 25 mL</td>
<td>00686</td>
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<tr>
<td>1</td>
<td>Funnel, long stem</td>
<td>00179</td>
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<td>Large test tube, 25 x 150 mm</td>
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<td>1</td>
<td>Test tube rack</td>
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<td>1</td>
<td>Test tube holder</td>
<td>00106</td>
<td>$3.18</td>
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<tr>
<td>1</td>
<td>Spatula, nichrome</td>
<td>00233</td>
<td>$4.64</td>
</tr>
<tr>
<td>2</td>
<td>Stirring rods, glass</td>
<td>00244</td>
<td>$0.42</td>
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<tr>
<td>1</td>
<td>Watchglass, 3 &quot; diameter</td>
<td>01101</td>
<td>$3.23</td>
</tr>
<tr>
<td>1</td>
<td>Evaporating dish, ceramic</td>
<td>00715</td>
<td>$5.13</td>
</tr>
<tr>
<td>1</td>
<td>Wash bottle, 250 mL, plastic</td>
<td>00584</td>
<td>$2.33</td>
</tr>
<tr>
<td>1</td>
<td>Storage bottle w/ cap, plastic</td>
<td>00582</td>
<td>$2.10</td>
</tr>
</tbody>
</table>

**NOTE:** If your lab drawer contains any of the following equipment, please return the items to the collection bins provided.

2 Medicine droppers  
1 Crucible bottom, ceramic  
1 Crucible cover, ceramic  
1 Triangular pipestem  
1 Wire gauze square  
1 Tongs, crucible

**PREPARATION:** Meaningful lab experiences require pre-laboratory preparation. Refer to the experiment schedule on the last page of this syllabus to determine which experiments will be performed each week. A scientist ordinarily begins with a literature search. You should begin by reading the introductory information provided in each experiment. Additional reading may be assigned within the introductory material or by your instructor.

You must complete the prelab assignment before coming to lab; otherwise you will not be permitted to do the experiment and will lose points. The prelab will help clarify your understanding of the reading. If you have questions concerning the prelab, seek help by stopping by Rm 508 at one of the course office hours. When you arrive to lab, turn in your prelab immediately, and then check your answers with the posted key. The answer key is a useful tool if used properly. You may use it to check your answers, but not to get your answers.
Read through the experimental procedure before coming to lab. This will help you use your time more efficiently in the lab and minimize the frustration associated with not knowing what is going on. The experiments can be completed in the allotted amount of time if you come to lab prepared.

It is important that you arrive on time. Your instructor will give a brief introduction before you begin to work, noting key points about the experiment and special safety considerations. Students who arrive after the introduction begins are not permitted to do the experiment and will be required to schedule a makeup session in order to do so.

Remember that some experiments must be extremely precise while others need only be approximate. Scientists and engineers must always keep in mind that unnecessarily careful measurement can steal time from other work. On the other hand, the results of rough measurement can be misleading. It is important, therefore, to select the correct equipment and operate it skillfully to have a successful outcome.

University chemistry laboratory does not require a high degree of mathematical sophistication, but reasonable skill in problem solving is necessary to complete the calculations and reports. The math skills required include: algebraic manipulations, dimensional analysis (factor-label method), manipulation of exponents, scientific notation, graphing and significant figures. You should review these topics and seek assistance as needed.

REPORTS AND RECORD KEEPING: Each experiment has a report form. Accurate reporting of experimental results is very important in laboratory work. Responses should be thorough and clearly written. If you were to refer to your report months later, you should be able to remember what you did, accurately. Others reading your report should be able to determine what you did. Points will be deducted from reports that fail to comply with the following criteria:

- Completed prelab assignments are due at the beginning of each lab session, before your instructor begins the introduction, and will be returned to you with your completed report the following session. After you have turned them in, check the prelab answer key to see if you were on the right track before you start the experiment.
- Data pages must be completely blank before you begin work.
- Reports must be written in permanent ink and must be written legibly.
- Record all data and observations at the time they are obtained. Tables are provided to help you organize your numerical data. Blank spaces are provided for observations. Record initial and final observations whether a change has occurred or not. All observations are important! Sometimes things that are not seen are as important as those that are.
- Data should be reported to the correct number of significant figures and include units, labels, and lead zeros (0.123, NOT .123).
- Never erase. Do not use White-out™ (liquid paper). Make corrections by drawing a single line through the erroneous data and entering the correct data next to it. Complete obliteration of the erroneous data is not acceptable. See the examples posted in the laboratory of acceptable and unacceptable work.
- Reports and postlabs are due the day the experiment is completed and must be turned in to your instructor before leaving the lab.
• Where calculations using data are involved, show one example of each type of calculation using data from your first trial unless noted otherwise. Always use units and labels in your calculations and pay attention to significant figures.

Collaboration/Plagiarism Rules

ACADEMIC INTEGRITY AND HONESTY: One of a scientist's (person's) most prized possessions is integrity. A scientist records all data and observations exactly as they occur, even if the unexpected is observed. Do not change data or observations to what you think they should be, but rather, try to explain the unexpected. Scientists learn by discussion with one another. You may also profit by discussion with your classmates, but not by copying from them. Cheating is considered a serious offense and is not tolerated in the lab. Examples of cheating for this course follow. The Office of Student Affairs (OSA) will be notified if a student is detected cheating. The OSA will take action consistent with procedures set forth in the MTU Academic Integrity policy. Possible sanctions include an academic integrity warning, failure of the course, suspension or expulsion.

• Copying raw data for a lab without actually participating in the work resulting in the raw data, or without the permission of the instructor in charge of the course.
• Inventing raw data.
• Filling in parts of lab reports that require the raw data for calculations or interpretation before the data is collected.
• Holding discussions so thorough that they result in identical methods and numbers for problems for lab reports and homework assignments.
• The use of old lab reports for anything other than formatting purposes without the permission of the instructor in charge of the course. (Because there is no need for formatting lab reports in this course, old lab reports are forbidden in the laboratory. Possession of old reports will be construed as intent to cheat.)
• Allowing anyone to copy a lab report, homework, assignment, quiz or test; either now or in the future, without the permission of the instructor in charge of the course. Except for those experiments where you work with a partner or in a group, you are expected to collect data and observations and answer questions independently. Even in a team experiment, your report should be prepared independently, in your own words.

University Policies (from http://www.mtu.edu/ctl/instructional-resources/syllabus/syllabus_policies.html)

Academic Integrity  http://www.admin.mtu.edu/usenate/policies/p109-1.htm: Academic regulations and procedures are governed by University policy. Academic misconduct cases will be handled in accordance with University’s policies.

Assessment  http://www.admin.mtu.edu/usenate/policies/p312-1.htm: Student work products (exams, essays, projects, etc.) may be used for the purposes of university, program, or course assessment. All work used for assessment purposes will not include any individual student identification. CH1151 is a General Education assessed course.

Disability Services  http://www.mtu.edu/deanofstudents/students/disability/: If you have a disability that could affect your performance in any class or that requires an accommodation under the Americans with Disabilities Act, please contact your instructor or Disability Services at 487-1494 as soon as possible so that appropriate arrangements can be made.

The Office of Institutional Equity  http://www.mtu.edu/equity: Michigan Technological University complies with all federal and state laws and regulations regarding discrimination, including the Americans with Disabilities Act of 1990. Michigan Tech has a policy of affording equal opportunity to all of its employees, students, applicants for employment, and applicants for admission without regard to race, religion, color, national origin, age, sex, sexual orientation, gender identity, height, weight, genetic information, or marital status, disabled veteran status, veteran status, or disability.

Veterans / Military  http://www.mtu.edu/registrar/students/veterans/: Veterans and active duty military personnel with special circumstances (e.g., upcoming deployments, drill requirements, disabilities) are welcomed and encouraged to communicate these, in advance if possible, to their instructor(s).

Equal Opportunity/Discrimination/Harassment  http://www.admin.mtu.edu/admin/boc/policy/ch5/: For other concerns about discrimination, you may contact your advisor, Chair/Dean of your academic unit, Dean of Students Office at 487-2212 or The Office of Institutional Equity and Inclusion at 487-3310.
### CH1151 Experiment & Quiz Schedule*, Spring 2016

<table>
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<tr>
<th>Week</th>
<th>Activity</th>
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| Week 2 (18-22 Jan) | Introduction, Safety, Check In  
• Note that all experimental handouts (with prelabs) will be available on Canvas by noon on Friday the week prior to when each experiment is scheduled. Students are required to print their own copies for use in lab. |
| Week 3 (25-29 Jan) | Techniques of Measurement & Observation *(TMO on Canvas)* |
| Week 4 (1-5 Feb) | No CH1151 labs *Winter Carnival Week, Enjoy!* |
| Week 5 (8-12 Feb) | Valence Electrons & Lewis Structures *(Lewis on Canvas)* |
| Week 6 (15-19 Feb) | Molecular Geometry *(MoGeo on Canvas)* |
| Week 7 (22-26 Feb) | Phase Diagram of t-Butyl Alcohol *(Phase on Canvas)* |
| Week 8 (29 Feb - 4 Mar) | Spectroscopy of Food Dyes *(Spec on Canvas)* and Quiz 1  
• Quiz 1 covers lab material through Phase Diagram of t-Butyl Alcohol; no makeups for Quiz 1 after 5 pm Thurs, 3 Mar  
• Spring Break Recess begins at 10 pm Fri, 4 Mar  
• Note that Daylight Saving Time begins on Sun, Mar 13 at 2 AM, at which time it becomes 3 AM |
| Week 9 (14-18 Mar) | Sequence of Copper Reactions *(SeqCu on Canvas)* |
| Week 10 (21-25 Mar) | Acidity of Fruit Juices *(Acidity on Canvas)*  
• Last day to drop with a grade of ‘W’ is Fri, 25 Mar by 5 pm; drops must be made in person at the Student Service Center, not via the web |
| Week 11 (28 Mar - 1 Apr) | Calorimetry *(Calor on Canvas)* |
| Week 12 (4-8 Apr) | H₂O₂ Decomposition and the Ideal Gas Law *(H₂O₂ on Canvas)* |
| Week 13 (11-15 Apr) | Factors Affecting Reaction Rates *(Factors on Canvas)* and Quiz 2  
• Quiz 2 covers week 8 through week 12 lab material; no makeups for Quiz 2 after 5 pm Thurs, 14 Apr  
• No makeups for Factors Affecting Reaction Rates after 3 pm Thurs, 14 Apr |
| Week 14 (18-22 Apr) | Course Survey, Clean up, & Check out  
• Noon sections arrive at 1 pm; 3 and 7 pm sections arrive at the usual time  
• $25 plus the cost of any broken/missing equipment will be billed to your tuition account if you fail to check out of your drawer by 5 pm Thurs, 21 Apr  
• A course survey will be conducted in lab. We also strongly encourage you to complete the online university evaluation for CH1151. |

* Changes to this experiment/quiz schedule may be made by the lab supervisor. Changes will be updated on the syllabus posted at the CH1151 Canvas (you will be notified if a change occurs).